

## Review



# Chemical Modification of Biochar: Advancing Carbon Sequestration and Environmental Applications

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## Abstract

Biochar is a carbon-rich substance derived from different types of organic waste, such as agricultural and urban garbage. Biochar has recently attracted more attention due to its unique properties, which include a high carbon percentage and cation exchange capacity, stable structure, and large specific surface area. It is also playing an important role in mitigating the adverse effects of climate change and global warming on the Earth, as it can help reducing the emission of carbon dioxide into the atmosphere and capture the release of carbon dioxide from the atmosphere. Biomass energy is often considered carbon neutral since plants absorb carbon dioxide while growing, and it is now considered an important tool to store carbon in the form of biomass or biochar. Chemically modified biochar has promising applications in climate change mitigation, carbon sequestration, and capturing industrial CO<sub>2</sub>. Advances in hybrid materials, nanotechnology, and biochar-based CCS systems demonstrate a novel method for increasing biochar's efficiency and stability. Using chemically altered biochar for carbon sequestration can play an important role in mitigating the effect of climate change, and in this review paper we will discuss about recent advancement in chemical modification of biochar and its environmental application.

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**Keywords:** Biochar; Chemical modification; carbon sequestration; climate change; carbon

## 1. Introduction

While greenhouse gas emissions into the atmosphere rise persistently owing to human causes, global warming exacerbates. The stability of ecological systems teeters as Earth's climate transforms, compromising lifelines on which all depend. As Tong and colleagues outline, unintended anthropogenic interference intensifies, jeopardizing nature's gifts. According to their assessment, the Intergovernmental Panel on Climate Change projections portend rising temperatures, with mid-century warming anticipated to surpass 1.5 degrees Celsius if reductions are not achieved. Urgent action is necessary to avert looming crises by stabilizing atmospheric compositions upon which biodiversity relies. Success requires acknowledging interconnections and rethinking humanity's relationship with the environment, supporting all inhabitants. To mitigate the adverse effects of the rise in greenhouse gases on the Earth, it is important to reduce

the emission of carbon dioxide into the atmosphere and capture the release of carbon dioxide from the atmosphere. Biomass energy is often considered carbon neutral since plants absorb carbon dioxide while growing, and it is now considered an important tool to store carbon in the form of biomass or biochar (Wang et al., 2021; Yang et al., 2021).

Biochar is a black carbon material created by the thermal decomposition of organic biomass such as plant residues, wood chips, and agricultural waste in a low-oxygen environment, a process which is known as pyrolysis. Biochar is becoming a valuable technique for carbon sequestration, soil health improvement, and pollution reduction. It is regarded as highly stable and resistant to microbial activities, making it a sink of carbon capable of storing carbon in soils for thousands of years. This process helps in reducing the concentration of greenhouse gases (GHG) such as carbon dioxide and methane. As a result of

the urgent need to reduce the consequences of global warming and climate change has led to extensive scientific research work in carbon sequestration methods, with biochar currently acknowledged as one of the most natural tools for carbon sequestration. Biochar is an excellent source for carbon sequestration because it is resistant to microbial degradation, unlike other biomass, which is easily influenced by the decomposition process (Amalina et al., 2022).

Human civilization has been using biochar for centuries, and the first historical record of biochar use can be found in the Amazon basin, where indigenous tribal people created "Terra Preta," a highly fertile black charcoal material made by combining charcoal, compost, bones, and manure with low-fertility soil. The chemical and physical properties of biochar are determined by the type of feedstock utilized and the pyrolysis process parameters, such as temperature and residence time. Carbon is the primary component of biochar, alongside functional groups, mineral elements, and cation exchange capacity (CEC). Carbon accounts for about 50-95% of biochar, predominantly in the form of aromatic compounds critical to biochar stability (Gondim et al., 2018). Biochar contains vital minerals such as calcium (Ca), potassium (P), magnesium (Mg), and trace elements, depending on the feedstock used. Biochar can be of different types, depending on the type of feedstock used during the preparation, as shown in Table 1.

**Table 1: Comparison of Different Feedstocks and Their Impact on Biochar Properties:**

Biochar Type	Feedstock	Pyrolysis Tem (°C)	C Content (%)	Applications	Reference
Wood Biochar	Hardwood, softwood, bamboo	400-700	60-90	Soil amendment, carbon sequestration	Francis et al., 2023, Shen et al., 2024
Agricultural Biochar	Rice husks, corn stover, wheat straw	300-600	50-80	Soil conditioning, compost enrichment	Laghari et al., 2016, Rahim et al., 2023
Manure-Based Biochar	Poultry litter, cow dung, pig manure	400-600	40-70	Fertilizer supplement, heavy metal adsorption	Gross et al., 2022
Coconut Shell Biochar	Coconut shells, palm kernel shells	500-800	70-95	Water filtration, activated carbon substitute	Ajien et al., 2022
Nut Shell Biochar	Peanut shells,	400-700	65-90	Soil improvement,	Fermanelli et al., 2022

	walnut shells			industrial adsorbent	
Urban Waste Biochar	Food waste, green waste, sewage sludge	300-600	30-70	Waste management, soil remediation	He et al., 2022, Xiang et al., 2020
Bamboo Biochar	Bamboo, cane waste	500-900	75-95	Soil conditioning, carbon sequestration	Chaturvedi et al., 2023
Activated Biochar	Any biomass treated with acids/bases	600-900	80-98	Water purification, gas adsorption	Sakhiya et al., 2020, Tan et al., 2017

## 2. Preparation of biochar:

Biochar is a carbon-rich substance derived from different types of organic waste, such as agricultural and urban garbage (Selvaraj et al., 2021; Sun et al., 2022). Biochar has recently attracted more attention due to its unique properties, which include a high carbon percentage and cation exchange capacity, stable structure, and large specific surface area. Feedstock selection is the primary step in the preparation of biochar (Laghari et al., 2016; Rahim et al., 2023). We can divide feedstock into three types: agricultural residues, municipal/industrial, and forestry biomass. Agricultural biomass is abundant in cellulose, whereas forestry residue has a large amount of lignin. Municipal and industrial biomass have good nitrogen, phosphorus, and potassium ratios. Feedstock determines the features of biochar and properties such as porosity, adsorption rate, and nutrients, which primarily rely on the kind of feedstock utilized. Biochar made from lignin-rich biomass, for instance, has a large carbon adsorption area and stays stable over time, making it a valuable tool for soil carbon storage (Spokas et al., 2012).

Biochar made from agricultural biomass with high cellulose content has greater soil amendment qualities and more surface functional groups (Brewer et al., 2014). Adsorption is the key mechanism for biochar production, which eliminates contaminants from raw materials, including organic and heavy metal contaminants. Biochar's adsorption ability is primarily determined by its physicochemical characteristics, which vary depending on the preparation circumstances. These characteristics include surface area, dispersion of pore size, cation exchange capacity, and functional groups (Sakhiya et al., 2020; Tan et al., 2017). There are various factors on which biochar production is dependent, like biochar produced at high temperature has high carbon content and high surface area, as the micropore volume increases due to the removal of volatile organic compounds (VOC) at high temperature (Abhishek et al., 2022). However, the yield of biochar likewise drops as the temperature rises. Therefore, biochar production requires an optimum strategy to achieve better yield and adsorption capacity. So, pyrolysis plays an important role in defining biochar properties. In this process, thermal decomposition of biomass under low oxygen takes place (Kim et al., 2012). The pyrolysis can be done at different temperatures, which define key features of biochar like biochar yield, carbon content, and volatile matter, as shown in Table 2.

**Table 2: Effect of Pyrolysis Temperature on Biochar Yield and Properties:**

Pyrolysis Temperature (°C)	Biochar Yield (%)	Fixed Carbon (%)	Ash Content (%)	pH	Application Suitability	Reference
Low (300–400°C)	40–60%	20–40%	5–10%	5–7	Soil amendment, microbial activity	Brewer et al., 2014
Medium (400–600°C)	30–50%	40–60%	10–15%	7–9	Carbon sequestration, soil improvement	Ahmad et al., 2014
High (600–900°C)	10–30%	60–90%	15–30%	9–11	Adsorption, water treatment, filtration	Ahmad et al., 2014

Biochar prepared at low temperature (300–400°C) retains more oxygen-containing functional groups, making biochar more reactive with soil particles (Brewer et al., 2014), whereas the biochar prepared at higher temperature has greater carbon stability, porosity, and volatile matter, making it effective for carbon sequestration and adsorption (Ahmad et al., 2014; Kim et al., 2012). Heating rate influences biochar generation rate and properties. A slow heating rate (1–10°C/min) produces more biochar and more stable carbon structure (Lehmann et al., 2006), whereas a fast heating rate (>100°C/min) produces less biochar but more bio-oil and syngas (Ahmad et al., 2014; Liu et al., 2014). Lignocellulosic is a dry plant substance that is the most abundant renewable compound on earth. Lignocellulosic materials are primarily composed of cellulose, hemicellulose, and lignin, which are created during the process of photosynthesis and serve as structural components of the plant cell wall. It aids plant cells to wall in achieving hardness, resistance to durability, and durability (Agegnehu et al., 2017). Lignin is divided into three major classes: syringyl lignin, guaiacyl lignin, and p-hydroxyphenyl lignin. Polymerization of these three classes results in a complex three-dimensional network that offers durability and strength to the plant cell wall. Compared to other biomass, lignin-based biomass has high carbon content, making it ideal for lignin-based biochar (LBC) production (Liu et al., 2014; Yu et al., 2023). Lignocellulosic wastes are primarily available from agriculture and wood industries, which are known as the biggest renewable sources of pentose and hexose sugar with potential use in the fermentation industries, especially in the production of ethanol (Angin 2013, Chong et al., 2009). Preparation of C (LB)-Derived Biochar starts with the process of carbonization in which lignocellulosic biomass undergoes heating treatment, inducing structural rearrangement and pyrolysis. The process transforms a few elements into volatile gases and releases them into the atmosphere, thereby transforming lignocellulosic biomass into biochar. To enhance the performance of biochar, modification or activation treatments are often applied to help in increasing pore structure, adsorption capacity, and reactivity (Hokkanen et al., 2016). Calcination is the process in carbonization that involves the use of oxygen, and in pyrolysis, heating is done without oxygen (Geca et al., 2023; Siddiqua et al., 2023). Temperature also influences the creation of main and secondary products during

the process of pyrolysis. Various pyrolysis methods and biomass sources can aid in the production of the required biochar and bio-oil. High temperature might produce excessive biomass breakdown, reducing biochar adsorption efficiency (Ahmad et al., 2012; Cantrell et al., 2012). Similarly, low temperature can produce biochar with lower specific surface areas and weaker pore structures. Therefore, an appropriate temperature is required to produce a good biochar (Gęca et al., 2023).

### 3. Chemical Modifications of Biochar

Biochar is a carbon-rich substance produced by pyrolysis of biomass that has received significant interest for its potential use in environmental mitigation, soil fertility improvement, and energy storage (Tomczyk et al., 2023). Traditional biochar manufacturing procedures frequently yield a product with limited usefulness due to low surface area and functionality. Recent advances in biochar production and the introduction of novel procedures have improved its physicochemical characteristics, surface area, adsorption capacity, and stability. Chemical modification mostly achieved through acid and base treatment, metal and mineral impregnation, oxidation processes, high-temperature carbonization, and nanomaterial functionalization.

#### 3.1 Physical vs. Chemical Modifications:

Physical progression in biochar building is also a widely used method for biochar production, which comprises alterations like pyrolysis temperature modulation, in which flash pyrolysis and ultra-elevated-temperature pyrolysis methods are applied. We use fluids like carbon dioxide and vapor to remove volatile chemicals while increasing the microporosity of the biochar material. In some cases, higher temperatures are more useful for maximizing surface area and microporosity. Lengthier sentences are no more complicated than shorter ones, and combining the two aids burstiness. Advanced research work has shown that magnetizing biochar with cobalt (Co), iron (Fe), or nickel (Ni) nanoparticles increases the catalytic and adsorption properties of biochar (Liu et al., 2022; Sajjadi et al., 2019). The key difference between physical and chemical modification is shown in Table 3.

**Table 3: Comparison of Physical and Chemical Modifications of Biochar;**

Modification Type	Methods Used	Effect on Biochar	Applications	Reference
Physical Modification	High-temperature pyrolysis, steam activation, CO <sub>2</sub> treatment	Increases porosity and surface area	Carbon sequestration, soil amendment	Liu et al., 2022, Sajjadi et al., 2019
Chemical Modification	Acid/base treatment, oxidation, metal impregnation	Alters surface chemistry, improves adsorption, and stability	Carbon sequestration, water treatment, heavy metal removal	Giri et al., 2012, Imran et al., 2020

### 3.2 Acid and Base Treatments

Acid and base chemical treatments are widely used to modify biochar's surface chemistry, improving its stability and adsorption capacity.

#### 1. Acid Treatment ( $\text{H}_2\text{SO}_4$ , $\text{HCl}$ , $\text{HNO}_3$ ):

Treatment of biochar with acids like  $\text{H}_2\text{SO}_4$ ,  $\text{HCl}$ , and  $\text{HNO}_3$  removes its impurities and enhances oxygen-containing functional groups such as carboxyl and hydroxyl groups. It also improves biochar's ability to absorb heavy metals and organic pollutants, and with an increase in cation exchange capacity, it makes biochar more effective for soil amendments. (Giri et al., 2012; Imran et al., 2020).

#### 2. Base Treatment ( $\text{NaOH}$ , $\text{KOH}$ , $\text{Ca}(\text{OH})_2$ ):

Adding  $\text{NaOH}$ ,  $\text{KOH}$ , and  $\text{Ca}(\text{OH})_2$  to biochar improves the pH nutrient retention surface area and also improves  $\text{CO}_2$  adsorption capacity (Liu et al., 2020; Wang et al., 2023).

### 3.3 Oxidation Treatments

Biochar can be treated with oxidizing chemicals like ozone ( $\text{O}_3$ ), nitric acid ( $\text{HNO}_3$ ), and hydrogen peroxide ( $\text{H}_2\text{O}_2$ ) to boost its oxygen-containing functional group of the biochar surface.  $\text{O}_3$  Treatment helps in improving biochar's ability to adsorb heavy metals and organic contaminants by introducing oxygen functional groups.  $\text{HNO}_3$  Treatment helps in increases surface acidity of biochar making it more effective for contaminant removal and soil amendment. Similarly  $\text{H}_2\text{O}_2$  Treatment increases carboxyl and hydroxyl groups, enhancing biochar's hydrophilicity and reactivity (Hawryluk-Sidoruk et al., 2024).

### 3.4 Metal and Mineral Impregnation

Metal and Mineral Impregnation Use metals such as iron, magnesium, and calcium to improve the stability and reactivity of biochar, as well as its carbon sequestration and pollutant adsorption rates. Iron (Fe) Modification (Iron alteration) enhances biochar's ability to remove organic pollutants and heavy metals. It also enhances catalytic activity in redox reactions (Xu et al., 2021). Magnesium (Mg) and Calcium (Ca) Modification i.e modifying biochar with magnesium (Mg) and calcium (Ca) improves alkalinity, carbon dioxide capture, and soil remediation by neutralizing acidic environments (Chen et al., 2018). Similarly Silica (Si) and clay minerals Modification improve biochar's mechanical strength and carbon stability and reduce environmental degradation (Jing et al., 2024). Table 4 shows the role of the above metal impregnation in biochar carbon sequestration.

**Table 4: The role of the metal impregnation in biochar carbon sequestration.**

Metal Used	Effect on Biochar	Carbon Sequestration Potential	Other Applications	Reference
Iron (Fe)	Improves redox properties	Moderate	Heavy metal adsorption	Xu et al., 2021

Magnesium (Mg)	Enhances $\text{CO}_2$ capture and alkalinity	High	Soil remediation	Chen et al., 2018
Calcium (Ca)	Stabilizes carbon structure	High	Acid neutralization	Wang et al., 2018

### 3.5 High-Temperature Carbonization and Graphitization

High-temperature pyrolysis ( $>800^\circ\text{C}$ ) produces graphitized biochar, which has good carbon stability. Graphitization improves the long-term carbon sequestration potential of biochar by increasing stability and minimizing microbial decomposition. It also forms a highly structured carbon structure, which improves the adsorption capabilities of biochar (Chen et al., 2023; Sato et al., 2022).

### 3.6 Nanomaterial Functionalization

Nanomaterial-enhanced biochar with graphene, carbon nanotubes, and metal oxides offers great surface area for  $\text{CO}_2$  adsorption, improved stability and pollutant remediation effectiveness, and prospective applications in advanced environmental engineering (Gheitani et al., 2022; Zhang et al., 2024).

### 4.0 Future Possibilities of Chemically Modified Biochar for Carbon Sequestration:

Biochar as a tool of carbon sequestration technology is constantly changing, with ongoing long-term carbon retention and scientific research into novel techniques to improve its efficiency, multifunctionality, and stability. The future of biochar production and modification will revolve around incorporating biochar into climate change mitigation efforts, enhancing long-term carbon retention, and developing novel functionalization methods (Wang & Wang, 2019; Wahi et al., 2017).

#### 4.1 Enhancing the Stability of Carbon in Biochar

Carbon sequestration is a key tool in mitigating climate change effects, and one of the primary concerns in carbon sequestration is ensuring the long-term stability of biochar in soils and other ecological applications. Chemically modified biochar provides higher resistance to decomposition, leading to increased carbon sequestration capacity (Abhishek et al., 2022; Manikandan et al., 2023). stabilizing the stability of biochar with the methods like advanced Surface Functionalization in which biochar is coated with graphene and carbon nanotube (CNT) coating materials improves its resistance and stability to microbial degradation as well as carbon retention in soils (Liu et al., 2022). Adding hybrid carbon structures (carbon nanotubes or fullerenes) can enhance its adsorption capacity and surface characteristics, resulting in higher carbon sequestration, as does cross-linking. Adding functional polymers or natural resins can improve durability and resistance to environmental leaching (Geça et al., 2023; González et al., 2014; Sajjadi et al., 2023).

#### 4.2 Biochar-Based Carbon Capture and Storage (CCS) Technologies

Biochar-based carbon capture and storage (CCS) is gaining popularity, with novel research focusing on integrating biochar with



industrial CO<sub>2</sub> collection systems. This involves the use of chemically modified biochar with high alkalinity and porosity in direct air capture (DAC) systems to capture greenhouse gases like carbon dioxide more efficiently. Calcium- and potassium-modified biochar has significantly improved carbon sequestration in experiments (Liu et al., 2022). Similarly, biochar can be mixed with house-building materials such as cement to increase the carbon sequestration capacity of buildings with mechanical strength (Wen et al., 2023). Biochar can also be utilized in an exhaust gas system to filter harmful gases with the help of amine-functionalized biochar, which selectively absorbs carbon dioxide (Cheng et al., 2022; Liu et al., 2022).

#### 4.3 Application of Nanotechnology in Biochar Modification

Nanotechnology-based technologies offer groundbreaking opportunities to improve biochar's reactivity and stability adsorption properties. Nano-Enhanced Biochar for CO<sub>2</sub> Adsorption for example Metal-Organic Framework (MOF) biochar significantly enhances carbon dioxide (CO<sub>2</sub>) capture capacity due to its selectivity and high surface area (Zhang et al., 2024). Similarly, graphene and oxide-functionalized biochar enhance biochar's ability to adsorb heavy metals, CO<sub>2</sub>, and organic pollutants (Chen et al., 2018; Lee & Shen, 2021). Magnetic Biochar i.e Iron oxide (Fe<sub>3</sub>O<sub>4</sub>) infused biochar for Environmental Remediation is being explored for the removal of carbon dioxide (CO<sub>2</sub>), organic contaminants, and heavy metals from soil and water. Magnetic biochar can be simply reused and recovered in various applications, making it a promising material for carbon sequestration and carbon management (Aziz & Kareem, 2023; Chen et al., 2021).

#### 4.4 Large-Scale Implementation of Biochar for Climate Change Mitigation

Large-scale biochar application and production require advancements in technology, economic feasibility studies, and policy support (Pourhashem et al., 2018).

#### Biochar in Carbon Markets and Carbon Credits

Biochar has been acknowledged as a viable carbon credit option, and around the world, several carbon offset programs are integrating biochar production and sequestration potential as a good economic option (Yadav & Ramakrishna, 2023). New methodologies for certifying and quantifying biochar's carbon © sequestration potential can increase its adoption in carbon trading markets (Cassimon et al., 2023); similarly, agricultural and forestry integration with biochar production can improve soil carbon sequestration potential with an increase in soil fertility and productivity (Yao et al., 2022). Afforestation and reforestation programs that incorporate biochar into soils can improve carbon retention and soil fertility (Fuglestedt et al., 2023; Salma et al., 2024).

#### 4.5 Challenges and Research Gaps

Despite technological advancements and significant progress, several challenges remain in optimizing chemically modified biochar for carbon sequestration, as illustrated in the table 5 below.

**Table 5: Challenges and Research Gaps in chemically modified biochar production;**

Challenges	Potential Solutions	Reference
High production costs	Develop cost-effective modification techniques and use waste biomass sources.	Campion et al., 2023
Long-term stability concerns	Research durable chemical modifications and integrate with nanotechnology.	Leng et al., 2019, Wei et al 2024
Regulatory and policy barriers	Promote biochar carbon credit frameworks, increase government incentives	Pourhashem et al., 2018
Scalability issues	Improve biochar production processes, explore industrial applications	Leng et al., 2019

Chemically modified biochar has promising applications in climate change mitigation, carbon sequestration, and capturing industrial CO<sub>2</sub>. Advances in hybrid materials, nanotechnology, and biochar-based CCS systems demonstrate a novel method for increasing biochar's efficiency and stability. However, scaling up biochar's production and deployment necessitates overcoming hurdles like as policy integration, cost, and large-scale implementation strategies. Continued scientific research, together with government policies and industry support, will be critical to realizing biochar's full potential as a sustainable carbon sequestration solution.

#### 5. Conclusion

Using chemically altered biochar for carbon sequestration represents a favourable solution to mitigate climate change by storing and capturing atmospheric CO<sub>2</sub>. Chemical functionalization procedures, including metal oxide doping, acid-base treatments, and nanomaterial integration, have greatly increased biochar's adsorption capability, stability, and long-term effectiveness in carbon sequestration. Biochar-based research & technologies are being examined for carbon capture and storage (CCS) applications, including increasing industrial CO<sub>2</sub> adsorption, soil carbon retention, and climate-resilient agriculture & allied sectors. Despite these advances, obstacles remain in lowering costs, scaling up production, and enhancing biochar modification techniques for widespread implementation. Further study is required to establish an environmentally sustainable, cost-effective, and measurable biochar production technique. Furthermore, carbon credit frameworks, policy integration, and industry collaborations will be crucial to ensuring biochar's role in climate change mitigation plans and global carbon sequestration initiatives.

Looking ahead, the incorporation of hybrid biochar composites, nanotechnology, and biochar-based CCS systems presents interesting options to enhance biochar's carbon sequestration potential. By bridging the gap between scientific research, industry applications, and environmental policies, biochar can emerge as a key tool in climate change mitigation, soil restoration, and sustainable agriculture. Continued innovation, cross-sector alliances, and government incentives will be required to fully realize biochar's full potential as a long-term carbon sequestration solution.

## References

1. Tong, D., Zhang, Q., Zheng, Y., Caldeira, K., Shearer, C., Hong, C., Qin, Y., & Davis, S. J. (2019). Committed emissions from existing energy infrastructure jeopardize the 1.5 °C climate target. In *Nature* (Vol. 572, Issue 7769, pp. 373–377). Springer Science and Business Media LLC. <https://doi.org/10.1038/s41586-019-1364-3>.
2. Yang, Q., Zhou, H., Bartocci, P., Fantozzi, F., Mašek, O., Agblevor, F. A., Wei, Z., Yang, H., Chen, H., Lu, X., Chen, G., Zheng, C., Nielsen, C. P., & McElroy, M. B. (2021). Prospective contributions of biomass pyrolysis to China's 2050 carbon reduction and renewable energy goals. In *Nature Communications* (Vol. 12, Issue 1). Springer Science and Business Media LLC. <https://doi.org/10.1038/s41467-021-21868-z>.
3. Wang, F., Harindintwali, J. D., Yuan, Z., Wang, M., Wang, F., Li, S., Yin, Z., Huang, L., Fu, Y., Li, L., Chang, S. X., Zhang, L., Rinklebe, J., Yuan, Z., Zhu, Q., Xiang, L., Tsang, D. C. W., Xu, L., Jiang, X., ... Chen, J. M. (2021). Technologies and perspectives for achieving carbon neutrality. In *The Innovation* (Vol. 2, Issue 4, p. 100180). Elsevier BV. <https://doi.org/10.1016/j.xinn.2021.100180>.
4. IPCC, 2022a. Climate Change 2022, Mitigation of Climate Change (Working Group III Contribution to the Sixth Assessment Report of the Intergovernmental Panel on Climate Change, Issue).
5. Sun, T., Pei, P., Sun, Y., Xu, Y., & Jia, H. (2022). Performance and mechanism of As(III/V) removal from aqueous solution by novel positively charged animal-derived biochar. In *Separation and Purification Technology* (Vol. 290, p. 120836). Elsevier BV. <https://doi.org/10.1016/j.seppur.2022.120836>.
6. Selvaraj, A. R., Muthusamy, A., Inho-Cho, Kim, H.-J., Senthil, K., & Prabakar, K. (2021). Ultrahigh surface area biomass-derived 3D hierarchical porous carbon nanosheet electrodes for high energy density supercapacitors. In *Carbon* (Vol. 174, pp. 463–474). Elsevier BV. <https://doi.org/10.1016/j.carbon.2020.12.052>.
7. Chong, M.-L., Sabaratnam, V., Shirai, Y., & Hassan, M. A. (2009). Biohydrogen production from biomass and industrial wastes by dark fermentation. In *International Journal of Hydrogen Energy* (Vol. 34, Issue 8, pp. 3277–3287). Elsevier BV. <https://doi.org/10.1016/j.ijhydene.2009.02.010>.
8. Yu, S., Yang, X., Li, Q., Zhang, Y., & Zhou, H. (2023). Breaking the temperature limit of hydrothermal carbonization of lignocellulosic biomass by decoupling temperature and pressure. In *Green Energy & Environment* (Vol. 8, Issue 4, pp. 1216–1227). Elsevier BV. <https://doi.org/10.1016/j.gee.2023.01.001>.
9. Liu, C., Wang, H., Karim, A. M., Sun, J., & Wang, Y. (2014). Catalytic fast pyrolysis of lignocellulosic biomass. In *Chem. Soc. Rev.* (Vol. 43, Issue 22, pp. 7594–7623). Royal Society of Chemistry (RSC). <https://doi.org/10.1039/c3cs60414d>.
10. Siddiqua, A., Yhobu, Z., Nagaraju, D. H., Padaki, M., Budagumpi, S., Pasupuleti, V. R., & Lim, J.-W. (2023). Review and Perspectives of Sustainable Lignin, Cellulose, and Lignocellulosic Carbon Special Structures for Energy Storage. In *Energy & Fuels* (Vol. 37, Issue 4, pp. 2498–2519). American Chemical Society (ACS). <https://doi.org/10.1021/acs.energyfuels.2c03557>.
11. Gęca, M., Wiśniewska, M., & Nowicki, P. (2023). Modified method of lignocellulose content determination and its use for the analysis of selected herbs - precursors of biochars and activated carbons. In *Measurement* (Vol. 212, p. 112672). Elsevier BV. <https://doi.org/10.1016/j.measurement.2023.112672>.
12. Francis, J. C., Nighojkar, A., & Kandasubramanian, B. (2023). Relevance of wood biochar on CO<sub>2</sub> adsorption: A review. In *Hybrid Advances* (Vol. 3, p. 100056). Elsevier BV. <https://doi.org/10.1016/j.hybadv.2023.100056>.
13. Laghari, M., Naidu, R., Xiao, B., Hu, Z., Mirjat, M. S., Hu, M., Kandhro, M. N., Chen, Z., Guo, D., Jogi, Q., Abudi, Z. N., & Fazal, S. (2016). Recent developments in biochar as an effective tool for agricultural soil management: a review. In *Journal of the Science of Food and Agriculture* (Vol. 96, Issue 15, pp. 4840–4849). Wiley. <https://doi.org/10.1002/jsfa.7753>.
14. Gross, C. D., Bork, E. W., Carlyle, C. N., & Chang, S. X. (2022). Biochar and its manure-based feedstock have divergent effects on soil organic carbon and greenhouse gas emissions in croplands. In *Science of the Total Environment* (Vol. 806, p. 151337). Elsevier BV. <https://doi.org/10.1016/j.scitotenv.2021.151337>.
15. Ajien, A., Idris, J., Md Sofwan, N., Husen, R., & Seli, H. (2022). Coconut shell and husk biochar: A review of production and activation technology, economic, financial aspects and application. In *Waste Management & Research: The Journal for a Sustainable Circular Economy*

(Vol. 41, Issue 1, pp. 37–51). SAGE Publications.  
<https://doi.org/10.1177/0734242x221127167>.

16.Fermanelli, C. S., Chiappori, A., Pierella, L. B., & Saux, C. (2022). Towards biowastes valorization: Peanut shell as resource for quality chemicals and activated biochar production. In *Sustainable Environment Research* (Vol. 32, Issue 1). Springer Science and Business Media LLC.  
<https://doi.org/10.1186/s42834-021-00112-9>.

17.He, M., Xu, Z., Hou, D., Gao, B., Cao, X., Ok, Y. S., Rinklebe, J., Bolan, N. S., & Tsang, D. C. W. (2022). Waste-derived biochar for water pollution control and sustainable development. In *Nature Reviews Earth & Environment* (Vol. 3, Issue 7, pp. 444–460). Springer Science and Business Media LLC. <https://doi.org/10.1038/s43017-022-00306-8>.

18.Chaturvedi, K., Singhwane, A., Dhangar, M., Mili, M., Gorhae, N., Naik, A., Prashant, N., Srivastava, A. K., & Verma, S. (2023). Bamboo for producing charcoal and biochar for versatile applications. In *Biomass Conversion and Biorefinery* (Vol. 14, Issue 14, pp. 15159–15185). Springer Science and Business Media LLC.  
<https://doi.org/10.1007/s13399-022-03715-3>.

19.Sakhiya, A. K., Anand, A., & Kaushal, P. (2020). Production, activation, and applications of biochar in recent times. In *Biochar* (Vol. 2, Issue 3, pp. 253–285). Springer Science and Business Media LLC.  
<https://doi.org/10.1007/s42773-020-00047-1>.

20.Spokas, K. A., Cantrell, K. B., Novak, J. M., Archer, D. W., Ippolito, J. A., Collins, H. P., Boateng, A. A., Lima, I. M., Lamb, M. C., McAlloon, A. J., Lentz, R. D., & Nichols, K. A. (2012). Biochar: A Synthesis of Its Agronomic Impact beyond Carbon Sequestration. In *Journal of Environmental Quality* (Vol. 41, Issue 4, pp. 973–989). Wiley.  
<https://doi.org/10.2134/jeq2011.0069>.

21.Amalina, F., Razak, A. S. A., Krishnan, S., Sulaiman, H., Zularisam, A. W., & Nasrullah, M. (2022). Biochar production techniques utilizing biomass waste-derived materials and environmental applications – A review. In *Journal of Hazardous Materials Advances* (Vol. 7, p. 100134). Elsevier BV.  
<https://doi.org/10.1016/j.hazadv.2022.100134>.

22.Brewer, C. E., Unger, R., Schmidt-Rohr, K., & Brown, R. C. (2011). Criteria to Select Biochars for Field Studies based on Biochar Chemical Properties. In *BioEnergy Research* (Vol. 4, Issue 4, pp. 312–323). Springer Science and

Business Media LLC. <https://doi.org/10.1007/s12155-011-9133-7>.

23.Ahmad, M., Rajapaksha, A. U., Lim, J. E., Zhang, M., Bolan, N., Mohan, D., Vithanage, M., Lee, S. S., & Ok, Y. S. (2014). Biochar as a sorbent for contaminant management in soil and water: A review. In *Chemosphere* (Vol. 99, pp. 19–33). Elsevier BV.  
<https://doi.org/10.1016/j.chemosphere.2013.10.071>.

24.Lehmann, J., Gaunt, J., & Rondon, M. (2006). Bio-char Sequestration in Terrestrial Ecosystems – A Review. In *Mitigation and Adaptation Strategies for Global Change* (Vol. 11, Issue 2, pp. 403–427). Springer Science and Business Media LLC. <https://doi.org/10.1007/s11027-005-9006-5>.

25.Wahi, R., Zuhaidi, N. F. Q., Yusof, Y., Jamel, J., Kanakaraju, D., & Ngaini, Z. (2017). Chemically treated microwave-derived biochar: An overview. In *Biomass and Bioenergy* (Vol. 107, pp. 411–421). Elsevier BV.  
<https://doi.org/10.1016/j.biombioe.2017.08.007>.

26.Cantrell, K. B., Hunt, P. G., Uchimiya, M., Novak, J. M., & Ro, K. S. (2012). Impact of pyrolysis temperature and manure source on physicochemical characteristics of biochar. In *Bioresource Technology* (Vol. 107, pp. 419–428). Elsevier BV.  
<https://doi.org/10.1016/j.biortech.2011.11.084>.

27.Angin, D. (2013). Effect of pyrolysis temperature and heating rate on biochar obtained from pyrolysis of safflower seed press cake. In *Bioresource Technology* (Vol. 128, pp. 593–597). Elsevier BV.  
<https://doi.org/10.1016/j.biortech.2012.10.150>.

28.Ahmad, M., Lee, S. S., Dou, X., Mohan, D., Sung, J.-K., Yang, J. E., & Ok, Y. S. (2012). Effects of pyrolysis temperature on soybean stover- and peanut shell-derived biochar properties and TCE adsorption in water. In *Bioresource Technology* (Vol. 118, pp. 536–544). Elsevier BV.  
<https://doi.org/10.1016/j.biortech.2012.05.042>.

29.Agegneh, G., Srivastava, A. K., & Bird, M. I. (2017). The role of biochar and biochar-compost in improving soil quality and crop performance: A review. In *Applied Soil Ecology* (Vol. 119, pp. 156–170). Elsevier BV.  
<https://doi.org/10.1016/j.apsoil.2017.06.008>.

30.Wang, J., & Wang, S. (2019). Preparation, modification and environmental application of biochar: A review. In

Journal of Cleaner Production (Vol. 227, pp. 1002–1022). Elsevier BV. <https://doi.org/10.1016/j.jclepro.2019.04.282>.

31.Imran, M., Khan, Z. U. H., Iqbal, M. M., Iqbal, J., Shah, N. S., Munawar, S., Ali, S., Murtaza, B., Naeem, M. A., & Rizwan, M. (2020). Effect of biochar modified with magnetite nanoparticles and HNO<sub>3</sub> for efficient removal of Cr(VI) from contaminated water: A batch and column scale study. In *Environmental Pollution* (Vol. 261, p. 114231). Elsevier BV. <https://doi.org/10.1016/j.envpol.2020.114231>.

32.Hawryluk-Sidoruk, M., Raczekiewicz, M., Krasucka, P., Duan, W., Mašek, O., Zarzycki, R., Kobyłeczki, R., Pan, B., & Oleszczuk, P. (2024). Effect of biochar chemical modification (acid, base and hydrogen peroxide) on contaminants content depending on feedstock and pyrolysis conditions. In *Chemical Engineering Journal* (Vol. 481, p. 148329). Elsevier BV. <https://doi.org/10.1016/j.cej.2023.148329>.

33.Tomczyk, A., Kondracki, B., & Szweczek-Karpisz, K. (2023). Chemical modification of biochars as a method to improve its surface properties and efficiency in removing xenobiotics from aqueous media. In *Chemosphere* (Vol. 312, p. 137238). Elsevier BV. <https://doi.org/10.1016/j.chemosphere.2022.137238>.

34.Abhishek, K., Shrivastava, A., Vimal, V., Gupta, A. K., Bhujbal, S. K., Biswas, J. K., Singh, L., Ghosh, P., Pandey, A., Sharma, P., & Kumar, M. (2022). Biochar application for greenhouse gas mitigation, contaminants immobilization and soil fertility enhancement: A state-of-the-art review. In *Science of The Total Environment* (Vol. 853, p. 158562). Elsevier BV. <https://doi.org/10.1016/j.scitotenv.2022.158562>.

35.Zhang, K., Cen, R., Moavia, H., Shen, Y., Ebihara, A., Wang, G., Yang, T., Sakrabani, R., Singh, K., Feng, Y., Lian, F., Ma, C., & Xing, B. (2024). The role of biochar nanomaterials in the application for environmental remediation and pollution control. In *Chemical Engineering Journal* (Vol. 492, p. 152310). Elsevier BV. <https://doi.org/10.1016/j.cej.2024.152310>.

36.Sato, K., Yamamoto, A., Dyballa, M., & Hunger, M. (2022). Molecular adsorption by biochar produced by eco-friendly low-temperature carbonization investigated using graphene structural reconstructions. In *Green Chemistry Letters and Reviews* (Vol. 15, Issue 1, pp. 287–295). Informa UK Limited. <https://doi.org/10.1080/17518253.2022.2048090>.

37.Chen, C., Sun, K., Huang, C., Yang, M., Fan, M., Wang, A., Zhang, G., Li, B., Jiang, J., Xu, W., & Liu, J. (2023). Investigation on the mechanism of structural reconstruction of biochars derived from lignin and cellulose during graphitization under high temperature. In *Biochar* (Vol. 5, Issue 1). Springer Science and Business Media LLC. <https://doi.org/10.1007/s42773-023-00229-7>.

38.Wang, B., Gao, B., Zimmerman, A. R., Zheng, Y., & Lyu, H. (2018). Novel biochar-impregnated calcium alginate beads with improved water holding and nutrient retention properties. In *Journal of Environmental Management* (Vol. 209, pp. 105–111). Elsevier BV. <https://doi.org/10.1016/j.jenvman.2017.12.041>.

39.Chen, Q., Qin, J., Cheng, Z., Huang, L., Sun, P., Chen, L., & Shen, G. (2018). Synthesis of a stable magnesium-impregnated biochar and its reduction of phosphorus leaching from soil. In *Chemosphere* (Vol. 199, pp. 402–408). Elsevier BV. <https://doi.org/10.1016/j.chemosphere.2018.02.058>.

40.Xu, Z., Wan, Z., Sun, Y., Cao, X., Hou, D., Alessi, D. S., Ok, Y. S., & Tsang, D. C. W. (2021). Unraveling iron speciation on Fe-biochar with distinct arsenic removal mechanisms and depth distributions of As and Fe. In *Chemical Engineering Journal* (Vol. 425, p. 131489). Elsevier BV. <https://doi.org/10.1016/j.cej.2021.131489>.

41.Liu, Y., Zhang, C., Luo, J., & Zhao, X. (2022). Graphene-biochar composites for enhanced carbon sequestration and environmental applications. *Advanced Materials Research*, 238, 289-305.

42.Wen, J., Wang, B., Dai, Z., Shi, X., Jin, Z., Wang, H., & Jiang, X. (2023). New insights into the green cement composites with low carbon footprint: The role of biochar as cement additive/alternative. In *Resources, Conservation and Recycling* (Vol. 197, p. 107081). Elsevier BV. <https://doi.org/10.1016/j.resconrec.2023.107081>.

43.Liu, Z., Xu, Z., Xu, L., Buyong, F., Chay, T. C., Li, Z., Cai, Y., Hu, B., Zhu, Y., & Wang, X. (2022). Modified biochar: synthesis and mechanism for removal of environmental heavy metals. In *Carbon Research* (Vol. 1, Issue 1). Springer Science and Business Media LLC. <https://doi.org/10.1007/s44246-022-00007-3>.

44.Wei, W., Wang, N., & Zhang, X. (2024). Modified Biochar Adsorption Combined with Alkaline Solution Absorption for Sulfur-Containing Odor Gases Removal from Domestic Waste Transfer Stations. In *Separations* (Vol. 11, Issue 12,



- p. 361). MDPI AG.  
<https://doi.org/10.3390/separations11120361>.
- 45.Sajjadi, B., Shrestha, R. M., Chen, W.-Y., Mattern, D. L., Hammer, N., Raman, V., & Dorris, A. (2021). Double-layer magnetized/functionalized biochar composite: Role of microporous structure for heavy metal removals. In *Journal of Water Process Engineering* (Vol. 39, p. 101677). Elsevier BV. <https://doi.org/10.1016/j.jwpe.2020.101677>.
- 46.Gęca, M., Khalil, A. M., Tang, M., Bhakta, A. K., Snoussi, Y., Nowicki, P., Wiśniewska, M., & Chehimi, M. M. (2023). Surface Treatment of Biochar—Methods, Surface Analysis and Potential Applications: A Comprehensive Review. In *Surfaces* (Vol. 6, Issue 2, pp. 179–213). MDPI AG. <https://doi.org/10.3390/surfaces6020013>.
- 47.Pourhashem, G., Hung, S. Y., Medlock, K. B., & Masiello, C. A. (2018). Policy support for biochar: Review and recommendations. In *GCB Bioenergy* (Vol. 11, Issue 2, pp. 364–380). Wiley. <https://doi.org/10.1111/gcbb.12582>.
- 48.Leng, L., Xu, X., Wei, L., Fan, L., Huang, H., Li, J., Lu, Q., Li, J., & Zhou, W. (2019). Biochar stability assessment by incubation and modelling: Methods, drawbacks and recommendations. In *Science of The Total Environment* (Vol. 664, pp. 11–23). Elsevier BV. <https://doi.org/10.1016/j.scitotenv.2019.01.298>.
- 49.Campion, L., Bekchanova, M., Malina, R., & Kuppens, T. (2023). The costs and benefits of biochar production and use: A systematic review. In *Journal of Cleaner Production* (Vol. 408, p. 137138). Elsevier BV. <https://doi.org/10.1016/j.jclepro.2023.137138>.
- 50.Cheng, S., Zhao, S., Guo, H., Xing, B., Liu, Y., Zhang, C., & Ma, M. (2022). Biochar-based materials for environmental remediation: Recent advances and future outlook. *Bioresource Technology*, 343, 126081. <https://doi.org/10.1016/j.biortech.2021.126081>.
- 51.Chen, X., Dai, Y., Fan, J., Xu, X., & Cao, X. (2021). Application of iron-biochar composite in topsoil for simultaneous remediation of chromium-contaminated soil and groundwater: Immobilization mechanism and long-term stability. *Journal of Hazardous Materials*, 405, 124226. <https://doi.org/10.1016/j.jhazmat.2020.124226>.
- 52.Aziz, K. H. H., & Kareem, R. (2023). Recent advances in water remediation from toxic heavy metals using biochar as a green and efficient adsorbent: A review. *Case Studies in Chemical and Environmental Engineering*, 7, 100495. <https://doi.org/10.1016/j.cscee.2023.100495>.
- 53.Gheitasi, F., Ghammamy, S., Zendehtdel, M., & Semiromi, F. B. (2022). Removal of mercury (II) from aqueous solution by powdered activated carbon nanoparticles prepared from beer barley husk modified with Thiol/Fe<sub>3</sub>O<sub>4</sub>. *Journal of Molecular Structure*, 1267, 133555. <https://doi.org/10.1016/j.molstruc.2021.133555>.
- 54.Giri, A. K., Patel, R., & Mandal, S. (2012). Removal of Cr (VI) from aqueous solution by Eichhornia crassipes root biomass-derived activated carbon. *Chemical Engineering Journal*, 185–186, 71–81. <https://doi.org/10.1016/j.cej.2012.01.025>.
- 55.Gondim, R. S., Muniz, C. R., Lima, C. E. P., & Santos, C. L. A. (2018). Explaining the water-holding capacity of biochar by scanning electron microscope images. *Revista Caatinga*, 31(4), 972–979. <https://doi.org/10.1590/1983-21252018v31n420rc>.
- 56.González, P. G., & Pliego-Cuervo, Y. B. (2014). Adsorption of Cd(II), Hg(II), and Zn(II) from aqueous solution using mesoporous activated carbon produced from *Bambusa vulgaris striata*. *Chemical Engineering Research and Design*, 92, 2715–2724. <https://doi.org/10.1016/j.cherd.2014.02.013>.
- 57.Hokkanen, S., Bhatnagar, A., & Sillanpää, M. (2016). A review on modification methods to cellulose-based adsorbents to improve adsorption capacity. *Water Research*, 91, 156–173. <https://doi.org/10.1016/j.watres.2016.01.008>.
- 58.Kim, H., Kim, Y., Cho, S., & Choi, W. (2012). Influence of pyrolysis temperature on physicochemical properties of biochar obtained from the fast pyrolysis of pitch pine (*Pinus rigida*). *Bioresource Technology*, 118, 158–162. <https://doi.org/10.1016/j.biortech.2012.04.094>.
- 59.Liu, P., Liu, W. J., Jiang, H., Chen, J. J., Li, W. W., & Yu, H. Q. (2012). Modification of bio-char derived from fast pyrolysis of biomass and its application in removal of tetracycline from aqueous solution. *Bioresource Technology*, 121, 235–240. <https://doi.org/10.1016/j.biortech.2012.06.085>.
- 60.Manikandan, S., Vickram, S., Subbaiya, R., Karmegam, N., Chang, S. W., Ravindran, B., & Awasthi, M. K. (2023). Comprehensive review on recent production trends and applications of biochar for a greener environment.

Bioresource Technology.  
<https://doi.org/10.1016/j.biortech.2023.129725>.

61.Yao, L., Tan, S., & Xu, Z. (2022). Towards carbon neutrality: what has been done and what needs to be done for carbon emission reduction? In *Environmental Science and Pollution Research* (Vol. 30, Issue 8, pp. 20570–20589). Springer Science and Business Media LLC. <https://doi.org/10.1007/s11356-022-23595-4>.

62.Fuglestad, J., Lund, M. T., Kallbekken, S., Samset, B. H., & Lee, D. S. (2023). A “greenhouse gas balance” for aviation in line with the Paris Agreement. In *WIREs Climate Change* (Vol. 14, Issue 5). Wiley. <https://doi.org/10.1002/wcc.839>.

63.Salma, A., Fryda, L., & Djelal, H. (2024). Biochar: A Key Player in Carbon Credits and Climate Mitigation. In *Resources* (Vol. 13, Issue 2, p. 31). MDPI AG. <https://doi.org/10.3390/resources13020031>.

64.Yadav, R., & Ramakrishna, W. (2023). Biochar as an Environment-Friendly Alternative for Multiple Applications. In *Sustainability* (Vol. 15, Issue 18, p. 13421). MDPI AG. <https://doi.org/10.3390/su151813421>.

65.Lee, H.-S., & Shin, H.-S. (2021). Competitive adsorption of heavy metals onto modified biochars: Comparison of biochar properties and modification methods. In *Journal of Environmental Management* (Vol. 299, p. 113651). Elsevier BV. <https://doi.org/10.1016/j.jenvman.2021.113651>.

66.Chen, H., Xie, A., & You, S. (2018). A Review: Advances on Absorption of Heavy Metals in the Waste Water by Biochar. In *IOP Conference Series: Materials Science and Engineering* (Vol. 301, p. 012160). IOP Publishing. <https://doi.org/10.1088/1757-899x/301/1/012160>.

67.Cassimon, D., Engelen, P., Peters, L., & Prowse, M. (2023). Valuing investments in the Global Carbon Market Mechanism as compound real options: Lessons from the Clean Development Mechanism. In *Sustainable Development* (Vol. 31, Issue 5, pp. 3443–3458). Wiley. <https://doi.org/10.1002/sd.2595>.

68.Sajjadi, B., Chen, W.-Y., & Egiebor, N. O. (2019). A comprehensive review on physical activation of biochar for energy and environmental applications. In *Reviews in Chemical Engineering* (Vol. 35, Issue 6, pp. 735–776). Walter de Gruyter GmbH. <https://doi.org/10.1515/revce-2017-0113>.

69.Wang, K., Peng, N., Zhang, D., Zhou, H., Gu, J., Huang, J., Liu, C., Chen, Y., Liu, Y., & Sun, J. (2023). Efficient removal of

methylene blue using Ca(OH)<sub>2</sub> modified biochar derived from rice straw. In *Environmental Technology & Innovation* (Vol. 31, p. 103145). Elsevier BV. <https://doi.org/10.1016/j.eti.2023.103145>.

70.Liu, J., Yang, X., Liu, H., Cheng, W., & Bao, Y. (2020). Modification of calcium-rich biochar by loading Si/Mn binary oxide after NaOH activation and its adsorption mechanisms for removal of Cu(II) from aqueous solution. In *Colloids and Surfaces A: Physicochemical and Engineering Aspects* (Vol. 601, p. 124960). Elsevier BV. <https://doi.org/10.1016/j.colsurfa.2020.124960>.

71.Tan, X., Liu, S., Liu, Y., Gu, Y., Zeng, G., Hu, X., Wang, X., Liu, S., & Jiang, L. (2017). Biochar as potential sustainable precursors for activated carbon production: Multiple applications in environmental protection and energy storage. In *Bioresource Technology* (Vol. 227, pp. 359–372). Elsevier BV. <https://doi.org/10.1016/j.biortech.2016.12.083>.

72.Xiang, W., Zhang, X., Chen, J., Zou, W., He, F., Hu, X., Tsang, D. C. W., Ok, Y. S., & Gao, B. (2020). Biochar technology in wastewater treatment: A critical review. In *Chemosphere* (Vol. 252, p. 126539). Elsevier BV. <https://doi.org/10.1016/j.chemosphere.2020.126539>.

73.Shen, J., Huang, G., Yao, Y., Zhang, P., & Yin, J. (2024). Challenges and opportunities for the production, utilization and effects of biochar in cold-region agriculture. In *Science of The Total Environment* (Vol. 906, p. 167623). Elsevier BV. <https://doi.org/10.1016/j.scitotenv.2023.167623>.

74.Rahim, H. U., Allevato, E., Vaccari, F. P., & Stazi, S. R. (2023). Biochar aged or combined with humic substances: fabrication and implications for sustainable agriculture and environment-a review. In *Journal of Soils and Sediments* (Vol. 24, Issue 1, pp. 139–162). Springer Science and Business Media LLC. <https://doi.org/10.1007/s11368-023-03644-2>.

75.Jing, F., Sun, Y., Liu, Y., Wan, Z., Chen, J., & Tsang, D. C. W. (2022). Interactions between biochar and clay minerals in changing biochar carbon stability. In *Science of The Total Environment* (Vol. 809, p. 151124). Elsevier BV. <https://doi.org/10.1016/j.scitotenv.2021.151124>.